

## SPATIAL VARIATION IN SOIL FERTILITY ALONG THE BANK OF HIRANYKESHI RIVER FROM GADHINGLAJ TAHSIL WITH REFERENCE TO SUGAR CANE FIELD

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**Abstract:** The present investigation deals with to evaluate the soil fertility status of Sugar Cane Field soil from bank of Hiranyakeshi River from Gadchinglaj Tahsil. Seven Soil samples were analyzed for various soil fertility parameters like pH. E.C., Available N, P, K, Sulphur, Calcium, Magnesium, Chloride, TDS and Water holding Capacity. During this study it has been found that there was a marked variation in nutrients levels at seven sites. It has also found that the only S<sub>4</sub> & S<sub>6</sub> soil samples Mg content below the normal range. All Macro and Micronutrients are within the normal range.

**Keywords:** Soil Parameter, Sugar Cane field, Soil Quality and Nutrients.

### Introduction

Soil fertility plays a key role in increasing crop production in the soil. It comprises not only in supply of nutrients but also their efficient Management. The fertility status of soil indicates their nutrient supplying capability. Moreover fertility of soil is subject to Man's control however soil degradation is an outcome depletive human activities and their interaction with natural environment (K.K. Deshmukh 2012). The evaluation of soil fertility includes the measurement of available plant nutrients and estimation of capacity of soil to maintain a continues supply of plant nutrients for a crop. The availability of nutrients depends upon various factors such on types of soils, nature of irrigation facilities and pH of organic matter content. Voluminous research works related to such aspects have been done by many researchers (S.K. Gupta et.al 1997). The three primary nutrients elements are N, P, and K also called as Micro nutrients which are not usually available in large enough amount for best growth. So they are most likely to be added to soil by fertilization. Soil commonly suffered from multiple nutrients deficiencies and nutrient imbalances, which leads to the decline in productivity and poor crop yield. So the need for an urgent means of assessing the fertility

status of these soils is necessary and the way to do so is by laboratory analysis of these soils (Sanda Ahmad R. et.al 2012).

**Study Area:** Gadhwaj Tahsil, Maps (Hiranyakeshi River)

### **Experimental**

Soil samples were collected from different location of sugar cane fields in and around the bank of Hiranyakeshi River. From the selective sites samples were collected in the depth of 20 cm from the surface of land which was taken in polythene bag as per the standard procedure (U.S. Salinity Laboratory Staff 1954, P.R. Hesse 1971). The samples were dried in air and passed through 2 mm sieve. The soil pH was measured by pH meter. The Potassium determination was carried out by Flame Photometer. The micronutrients were determined by Photometer in the Laboratory of Department of AGPM, Devchand College, Arjunnagar, Kagal Tahsil (M.S.). The dissolved element in the soil extract was measured by Atomic Absorption Spectrometer. Reagents used for the analysis were LR grade manufactured by S.D. Fine, Loba and Merck Fisher.

### **Result and Discussion**

For evaluation of soil fertility status of the study area pH, E.C., TDS, WHC, Available NPK, Sulphur, Calcium, Magnesium & Chloride were determined and results obtained were shown in Table no. 1.

**pH:** The soil pH value is measure of soil acidity or alkalinity and directly affect nutrients availability . The solubility of most nutrients varies in response to pH. As acidity increases the losses of these nutrients by leaching increases and their availability to plants decreases. The quantity of some nutrients may rise so greatly under acidic and alkaline conditions that they become toxic to plants (D. Briggs 1977). So soil pH is essential to control between 6.5 to 7.5. In the present study pH ranges from 6.7 to 8.5. The lower value of pH is recorded in the sample S<sub>1</sub> while it was higher in the remaining samples. These higher values are possibly due to presence of soluble and exchangeable Sodium along with HCO<sup>3-</sup>. High pH values are thus indicative of development of Salinity/ Sodicity in the area. Sample 1 shows acidic behavior.

### **Electrical Conductivity:**

The E.C. values ranges from 0.32 to 0.87 mhos. The higher value of E.C. have been seen in the sample S<sub>2</sub>, S<sub>4</sub> & S<sub>6</sub> while remaining samples shows lower values of Electrical Conductivity.

**Total Dissolved Solids:**

TDS for soil samples were ranges from 0.32 to 0.66. The higher value of TDS has been recorded in the sample S<sub>5</sub> & S<sub>6</sub>. The lowest values of TDS were recorded for sample S<sub>1</sub>, S<sub>2</sub>, S<sub>3</sub>, S<sub>4</sub> & S<sub>7</sub>.

**Water Holding Capacity:**

Sandy soil has lower water holding capacity as compared to clay soil. In present study water holding capacity values ranges from 0.37 to 0.61. The sample S<sub>1</sub>, S<sub>2</sub>, S<sub>3</sub> & S<sub>7</sub> has higher WHC value than sample S<sub>4</sub>, S<sub>5</sub> & S<sub>6</sub>.

**Nitrate Nitrogen:**

Nitrogen is an essential plant nutrient; it should be present in the soil in appropriate proportion for growth of plants. The plants absorbed Nitrogen either as Nitrate ion or Ammonium ion. The transformation of the Nitrogen compound in the soils involves the process such as fixation of atmospheric Nitrogen by free leaving and nodule forming bacteria considering unique importance of Nitrogen for the growth of plants attempt has been made to estimate available Nitrate Nitrogen and Ammonium Nitrogen from the soils in the study area.

Deficiency of Nitrogen is almost universal in Indian soil, so Nitrogen application is a must for canopy development and high yields. Reduced Nitrogen's lower the protein content of seeds and vegetative parts. In several cases flowering is greatly reduced Nitrogen deficiency causes early maturity in some crops which results in significant reduction in yield and quantity (J.A. Silva et.al 2000).

Available Nitrate Nitrogen in the soil sample ranged from 217.6 kg/ha to 257.92 kg/ha. Sample S<sub>6</sub> shows higher Nitrogen Nitrate value than the remaining samples.

Available Ammonium Nitrate in the soil samples ranged from 272 kg/ha to 409.9 kg/ha. Samples S<sub>3</sub>, S<sub>5</sub>, S<sub>7</sub> have higher Ammonium Nitrate values than the remaining.

**Potassium:**

Potassium available to plant as the ion K<sup>+</sup>, it assist in regulating the plants use of water by controlling the opening and the closing of leaf stomates, where water is released to cool the plant. Potassium has been shown to improve disease. Resistance in plant, improve the size of grains and seeds and improve the quality of fruit and vegetable.

Deficiency of Potassium reduced the size of seeds and fruits and the quantity of their production. In some crops stems are weak and lodging is common if K is deficient.

The available Potassium value in the soil ranges from 213.5 kg/ha to 303.5 kg/ha. It was high at sample S<sub>6</sub> & S<sub>1</sub> while soil sample S<sub>2</sub>, S<sub>3</sub>, S<sub>4</sub>, S<sub>5</sub>& S<sub>7</sub> was medium.

**Phosphorus:**

Phosphate ion enters the soil solution either as a result of mineralization of organophosphate or the application of fertilizers. The plant takes available P mostly in the form of  $H_2PO_4^-$  from soil solution. Chemisorptions of P occur due to interaction of Phosphate ion with the atoms like Al, Fe & Ca depending upon soil pH (D.S. Orlov 1992).

A phosphorus value ranges from 22.5 kg/ha to 50.16 kg/ha. Only three samples  $S_1$ ,  $S_2$  &  $S_7$  showed medium P content and soil sample  $S_3$ ,  $S_4$ ,  $S_5$  &  $S_6$  showed high P content value.

**Sulphur:**

Sulphur is an essential component in the synthesis of amino acids required to manufacture proteins. Sulphur is also required for production of Chlorophyll and utilization of Phosphorus and other essential nutrients. It increases the size and weight of grain crops and enhances the efficiency of Nitrogen for protein manufacture. It also increases quality of fiber crops.

An acute Sulphur deficiency causes the entire plant to turn yellow. In crops like corn and small grains, however yellow stripes that turn parallel to the leaf blade are common. Sulphur deficiency is most frequently observed on very sandy soils with low organic matter content during seasons of excessive rainfall (M. Ray Tucker 1999).

In the present study Sulphur content values ranges from 7.6 kg/ha to 19.2 kg/ha. Soil sample  $S_1$ ,  $S_2$ ,  $S_5$  &  $S_6$  has high Sulphur content as compared to  $S_3$ ,  $S_4$  &  $S_7$ .

**Calcium:**

Calcium is constituent of cell walls and is involved in production of new growing points and root tips. It provides elasticity and expansion of cell walls which keeps growing points from becoming rigid and brittle. It is immobile within plants and remains in the older tissue throughout the growing season. It acts as a base for neutralizing organic acids generated during the growing process and aids in Carbohydrate translocation and Nitrogen absorption.

Calcium deficiency symptoms appear in the new growth of leaves, stems, buds and roots. Younger leaves are affected first and are usually deformed. In extreme cases, the growing tips die. The leaves of some plants hook downward and exhibit marginal necrosis. Roots on Calcium deficiency plants are short and stubby.

In the present study Calcium content ranges from 11 ml/100 gm to 25 ml/100 gm. Sample  $S_1$ ,  $S_3$ ,  $S_4$ ,  $S_5$  &  $S_7$  having high Calcium content while sample  $S_2$  &  $S_6$  has low Calcium content. All samples were within the permissible limit.

### **Magnesium:**

Magnesium is required to stabilize ribosome particles and also helps to stabilize the structure of nucleic acid. It is available to plants as the ions Mg<sup>+2</sup>. Deficiency symptoms may occur in many plants when the leaf level is less than 0.10 to 0.15 %. Small grains may exhibit deficiency when the Mg level is less than 0.10 %. When the soil pH is less than 5.4 , Mg availability and uptake by plant is greatly reduced . Supplemental applications of fertilizer Mg may be needed in some cases to supply some of the Mg crop requirement (AESL Plant analysis Handbook).

In present investigation Mg content in the soil sample ranged from 2 to 35 ml/100 gm. Sample S<sub>1</sub>& S<sub>4</sub> having very low Mg content (below standard value), while sample S<sub>3</sub> having very high Mg content (above standard value). Remaining samples lies within the normal range.

### **Chloride:**

Chloride is essential in photosynthesis it increases cell osmotic pressure and water content of plant tissue (Shobha Jadhav 2012). Chloride deficiency mainly found in cereal crops (Mengel K et.al 2001).

In the present study Chloride content in the soil sample ranges from 25 to 50. All soil samples were in the normal range. Soil samples S<sub>1</sub>, S<sub>3</sub>, S<sub>4</sub>, S<sub>6</sub> and S<sub>7</sub> have same Chloride content (25 ml /100gm) and soil samples S<sub>2</sub> & S<sub>5</sub> having 50 ml/100gm Chloride content.

### **Conclusion:**

In the present investigation, the seven soil samples along the bank of Hiranyakeshi River from sugar cane field were collected. All Macronutrients and Micronutrients are within the normal range except Mg. Mg content in soil sample S4 and S6 below the normal range, so supplemental application of Mg containing fertilizer should be given. In case of pH of all soil samples are slightly alkaline in nature. All soil samples having Electrical conductivity values below one.

**Table 1:** Macronutrient and Micronutrients analysis of soil from Sugar cane field

| Soil Content | Hiralage S <sub>1</sub> | Mahagaon S <sub>2</sub> | Bhadgaon S <sub>3</sub> | Mugali S <sub>4</sub> | Nilagi S <sub>5</sub> | Khandal S <sub>6</sub> | Kadalage S <sub>7</sub> |
|--------------|-------------------------|-------------------------|-------------------------|-----------------------|-----------------------|------------------------|-------------------------|
| pH           | 6.7                     | 7.2                     | 8.5                     | 7.5                   | 7.5                   | 8                      | 7.9                     |
| TDS          | 0.47                    | 0.42                    | 0.32                    | 0.39                  | 0.61                  | 0.66                   | 0.38                    |
| EC           | 0.58                    | 0.87                    | 0.47                    | 0.64                  | 0.46                  | 0.86                   | 0.32                    |
| WHC          | 0.58                    | 0.57                    | 0.52                    | 0.47                  | 0.37                  | 0.42                   | 0.61                    |

|                     |        |       |        |       |        |       |       |
|---------------------|--------|-------|--------|-------|--------|-------|-------|
| Nitrate             | 228.91 | 217.6 | 241.92 | 219.7 | 221.76 | 257.6 | 217.6 |
| Nitrogen<br>(kg/ha) |        |       |        |       |        |       |       |
| Ammonium            | 272    | 339   | 409.9  | 272   | 303.07 | 272   | 406.5 |
| Nitrogen<br>(kg/ha) |        |       |        |       |        |       |       |
| Potassium           | 281    | 236   | 213.5  | 237   | 236.08 | 303.5 | 236   |
| (kg/ha)             |        |       |        |       |        |       |       |
| Sulphur             | 19.2   | 11.2  | 7.6    | 9.4   | 10.75  | 15.68 | 5.8   |
| (kg/ha)             |        |       |        |       |        |       |       |
| Calcium             | 25     | 15    | 25     | 30    | 20     | 11    | 19    |
| (ml/100kg)          |        |       |        |       |        |       |       |
| Magnesium           | 10     | 6     | 35     | 4     | 5      | 2     | 10    |
| (ml/100kg)          |        |       |        |       |        |       |       |
| Chloride            | 25     | 50    | 25     | 25    | 50     | 25    | 25    |
| (ml/100kg)          |        |       |        |       |        |       |       |
| Phosphorus          | 22.5   | 22.5  | 56     | 43.4  | 50.16  | 25.08 | 22.5  |
| (kg/ha)             |        |       |        |       |        |       |       |

## References

- [1] K.K. Deshmukh (2012). Evaluation of soil fertility status from Sangamner Area, Ahmednagar District, Maharashtra, India. *Rasayan Journal of Chemistry* 5(3): 398-406.
- [2] S.K. Gupta and I.C. Gupta. (1997). Management of Saline soil and water”, Scientific Publication Jodhpur.
- [3] Sanda Ahmad R. and Ismail Y.M. (2012). Soil fertility: Assessment of research and teaching farm of Audu Bakko College of Agriculture Danbatta, Kano State, Nigeria. *Nature Environment of Pollution Technology* 11(4): 639-642.
- [4] U.S. Salinity Laboratory Staff, Diagnosis and Improvement of Saline and Alkali soils USDA. Handbook Number 60, U.S. Department of Agriculture, Washington DC (1954).
- [5] P.R. Hesse. (1971).A textbook of soil chemical analysis, *John Murry Publication*, London U.K.

- [6] D. Briggs. (1977).Soil, Butterworth's and Co-publishers, London 129.
- [7] J.A. Silva and R. Uchida. (2000). Essential Nutrients for plant growth: Nutrient Functions and Deficiency Symptoms 31-55.
- [8] D.S. Orlov. (1992). Soil Chemistry, Oxford and IBH Publishers, New Delhi.
- [9] M. Ray Tucker. (1999). Agronomist: Essential plant Nutrients 1-9.
- [10] AESL Plant analysis Handbook, Nutrient content of plant. The University of Georgia, College of Agriculture and Environmental Sciences, Cooperative Extension Services. 1-5.
- [11] Shobha Jadhav (2012). Investigation of some parameters and Nutrients from Sewage irrigated and River water irrigated soil sample from Gadchinglaj, Maharashtra. *Rasayan Journal of Chemistry* 5(4): 493-495.
- [12] Mengel K. and E.A. Kirkby. (2001). Principles of Plant Nutrition Netherland, Kluwer Academic Publisher 849.