

## **GROUNDWATER QUALITY OF WELLS AND TUBE WELLS OF SUPAUL DISTRICT, BIHAR**

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**Abstract:** Water is must for life; the present piece of work is to study the status of the drinking water of Supaul district of Bihar. Ten tubewell and ten well water sample was collected from five villages of five villages of the district. The physico chemical parameters studied were pH, Electrical conductivity, free carbon dioxide, total alkalinity, total hardness, dissolved oxygen, Biological oxygen demand and chemical oxygen demand. The finding revealed that all the parameters are with permissible range of BIS except the COD which needs further study.

**Keywords:** Supaul, Drinking water, Physicochemical.

### **Introduction**

Water is one of the most important and abundant compounds of the ecosystem. But due to increased human population, industrialization, use of fertilizers in the agriculture and man-made activity it is highly polluted with different harmful contaminants. Therefore it is necessary that the quality of drinking water should be checked at regular time interval, because due to use of contaminated drinking water, human population suffers from varied of water borne diseases. It is difficult to understand the biological phenomenon fully because the chemistry of water reveals much about the metabolism of the ecosystem and explain the general hydro - biological relationship (Basavaraja Simpi et al. 2011).

The availability of good quality water is an indispensable feature for preventing diseases and improving quality of life. Natural water contains different types of impurities are introduced in to aquatic system by different ways such as weathering of rocks and leaching of soils, dissolution of aerosol particles from the atmosphere and from several human activities, including mining, processing and the use of metal based materials (Ipinmoroti and Oshodi 1993, Adeyeye 1994, Asaolu 1997). The increased use of metal-based fertilizer in agricultural revolution of the government could result in continued rise in concentration of metal pollutions in fresh water reservoir due to the water run-off.

Most of the rivers in the urban areas of the developing countries are the ends of effluents discharged from the industries. African countries and Asian countries experiencing rapid

industrial growth and this is making environmental conservation a difficult task (Agarwal Animesh 2011). Sea water contains large number of trace metals in very small concentration. This is a challenging matrix for the analytical chemist due to the very low concentrations of many important trace metals (Robertson 1968, Riley).

In north Bihar drinking water is a problem for the people throughout the year - during and after floods. During floods, the tube wells are submerged under water or covered up in silt, thus creating a problem of access.

The objective of the present study is to assess the physico-chemical properties of water collected from different sources.

### Study Area

The Supaul district is located at the north eastern parts of Bihar state, which is situated in the middle parts of Ganga Basin. The district falls in the Kosi Sub-basin.

In the district shallow tube wells are suitable up to a depth of 50 m with discharge of 20 to 40 m<sup>3</sup>/hr. The deep tube well of more than 100 m depth can also be constructed with estimated discharge of 100-200 m<sup>3</sup>/hr. Whereas Bamboo Boring of 20 to 25 m depth tapping water table aquifer can yield 10 to 20 m<sup>3</sup>/hr with a safe draw down.

Depth to water level in the Supaul district remains shallow during pre- as well as post-monsoon periods, going maximum up to 5.0 m at few patches.

### Sample collection site

| District | Block            | Village  | No. of samples |               |
|----------|------------------|----------|----------------|---------------|
|          |                  |          | Bore well      | Well          |
| Supaul   | <u>Chhatapur</u> | Amha     | 2 (BW1 & BW2)  | 2 (OW1 & OW2) |
|          | <u>Kishanpur</u> | Andauli  | 2 (BW3 & BW4)  | 2(OW3 & OW4)  |
|          | <u>Marauna</u>   | Barahara | 2(BW5 & BW6)   | 2(OW5 & OW6)  |
|          | <u>Nirmali</u>   | Bela     | 2(BW7 & BW8)   | 2(OW7 & OW8)  |
|          | Pipra            | Basaha   | 2 (BW9 & BW10) | 2(OW9 & OW10) |

### Materials and Methods

A total of 20 water samples of tube wells and wells were collected directly into 2.0 Ltr acid cleaned polythene bottles. These water samples were kept in the darkness in an ice box at 4<sup>0</sup>C till the samples reached the laboratory for analysis. The samples were analyzed for physicochemical using standard procedures (USEPA, 1990; APHA, 1992).

## **Results and Discussion**

The water quality analysis of 20 water samples has been carried out for 8 physico-chemical parameters, pH, Electrical conductivity, free carbon dioxide, total alkalinity, total hardness, dissolved oxygen, Biological oxygen demand and chemical oxygen demand.

### **pH**

The pH varied from 6.5-6.8 in tubewell water and 6.8-7.0 in open well water. pH is one of the importance on determining the corrosivity of water because generally the lower the pH, the higher the level of corrosion (WHO, 1996). Cautious attention to pH is necessary at all stages of water treatment before distribution to ensure satisfactory clarification and disinfection to minimize the corrosion of water. Exposure to extreme (pH > 11) results in irritation in eyes, skin and mucous membrane and also cause hair fibers to swell in human. Similarly, low pH also results in same effects with the severity of which increases with decreasing pH (WHO Working Group, 1986). According to the WHO guidelines, the taste of drinking water should be non-objectionable or acceptable to consumers. The taste also depends on the pH of water. The BIS of drinking water pH is 6.8 – 8. The pH of water sample of tube well and well is within range of BIS (BIS 2012)

### **Electrical conductivity (EC)**

Electrical conductivity is a measure of water capacity to convey electric current. It signifies the amount of total dissolved salts. EC values were observed in the range of 430.0 micromhos/cm to 490.0 micromhos/cm.in borewell water and 415.0 micromhos/cm to 450.0 micromhos/cm.in well water. EC values were found within WHO limit.

### **Free Carbon dioxide**

The main source of free carbon dioxide is respiration of the plant and animals and also the decomposition of the organic matter. It has been reported that minimum free carbon dioxide in water is good for health. In the present study the free carbon dioxide is ranged 6.0-6.4 ppm in tube well and 4.8-8.0 ppm in well water. There is no guide line of BIS for free carbon dioxide so it should be minimum in drinking water.

### **Total Alkalinity**

Total alkalinity of water may be due to the presence of one or more number of ions. These include hydroxides, carbonates and bicarbonates. Hydroxide ions are always present in water, even if the concentration is extremely low. However, significant concentrations of hydroxides are unusual in natural water supplies, but may be present after certain types of treatment. The

total alkalinity ranged from 152-190ppm in tube well water and 178-190ppm in well water. as per BIS it is 200-600 ppm. so the sample water is within range of BIS.

### **Total Hardness**

Total hardness observed for streams and rivers throughout the world range between 1-1000 ppm as CaCO<sub>3</sub>. Hardness reflects the composite measure of polyvalent cations whereas calcium and magnesium are the primary constituent of hardness (Larry, 1996). The measure value of total hardness for studies samples was 140172 ppm in tube well and 140180ppm in well water. The BIS recommendation is 200-600ppm. The sampled water the total hardness was found within the permissible limit of BIS.

### **Dissolved oxygen**

A high DO level in a community water supply is good because it makes drinking water taste better. However, high DO levels speed up corrosion in water pipes. The amount of dissolved oxygen often determines the number and types of organisms living in that body of water. the dissolved oxygen was 1.2-1.8ppm in tube well water and 2.4-4.2 ppm in well water. one the reason of high dissolved oxygen in well may the open well and direct contact with the air.

### **Biological oxygen demand (BOD)**

There is no specific limit prescribed for BOD in drinking water by BIS. The presence of BOD in drinking water indicates that the water contains biodegradable organic substances. The bacterial growth will start in such water very easily. The BOD should be Zero in drinking water, but it's almost impossible to maintain Zero BOD in open water source. The BOD level found 1.2-2.8 ppm in tube well and 4.2-4.6ppm in well water. BOD <5ppm will not cause any harmful impacts on human body, since the bacteria present in our digestive system will degrade all the organic contents.

### **Chemical Oxygen demand (COD)**

The COD in tube well water ranged 60-68 ppm and in well water 38.00-44.00 as per BIS standard for drinking water (IS 10500:1991), there is no mention of COD limit. These parameters are meant for Effluent samples not for drinking water. It means there should not be any trace of COD values for drinking water. The presence of COD in the water in area needs further detailed study about the reasons and remedies.

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**Table 1:** Physico Chemical status of Bore well and well of the project area

| parameters              | Village        | Amha |      | Andauli |      | Barahara |      | Bela |      | Basaha |      | Amha |      | Andauli |      | Barahara |      | Bela |      | Basaha |      |
|-------------------------|----------------|------|------|---------|------|----------|------|------|------|--------|------|------|------|---------|------|----------|------|------|------|--------|------|
|                         | Unit           | BW1  | BW2  | BW3     | BW4  | BW5      | BW6  | BW7  | BW8  | BW9    | BW10 | OW1  | OW2  | OW3     | OW4  | OW5      | OW6  | OW7  | OW8  | OW9    | OW10 |
| Temperature             | <sup>0</sup> C | 24.5 | 24.5 | 25.0    | 25.3 | 25.2     | 25.6 | 25.4 | 25.6 | 25.0   | 25.2 | 24.6 | 26.0 | 25.6    | 25.8 | 26.0     | 26.2 | 26.2 | 25.0 | 25.4   | 25.2 |
| pH                      |                | 6.5  | 6.6  | 6.8     | 6.8  | 6.4      | 6.8  | 6.7  | 6.8  | 6.8    | 6.8  | 7.0  | 6.8  | 6.8     | 6.8  | 6.8      | 6.8  | 6.8  | 6.8  | 6.6    | 6.8  |
| Electrical conductivity | mhos/cm        | 480  | 490  | 430     | 480  | 445      | 430  | 450  | 458  | 452    | 452  | 435  | 430  | 432     | 430  | 428      | 425  | 424  | 420  | 415    | 416  |
| Free CO <sub>2</sub>    | ppm            | 6.0  | 6.0  | 6.2     | 6.4  | 6.2      | 6.0  | 6.0  | 6.0  | 6.2    | 4.8  | 4.8  | 5.8  | 6.8     | 7.2  | 7.2      | 7.2  | 6.8  | 7.6  | 7.8    | 7.8  |
| Total Alkalinity        | ppm            | 156  | 152  | 158     | 180  | 182      | 189  | 190  | 190  | 168    | 176  | 184  | 190  | 182     | 178  | 182      | 184  | 186  | 182  | 182    | 180  |
| Total Hardness          |                | 140  | 152  | 154     | 146  | 142      | 146  | 148  | 158  | 170    | 172  | 142  | 158  | 172     | 175  | 174      | 180  | 182  | 140  | 146    | 144  |
| Dissolved Oxygen        | ppm            | 1.46 | 1.80 | 1.2     | 1.8  | 1.2      | 1.2  | 1.2  | 1.4  | 1.2    | 1.2  | 2.4  | 2.6  | 4.0     | 4.0  | 4.2      | 2.8  | 4.2  | 4.0  | 4.0    | 4.0  |
| BOD                     | ppm            | 1.2  | 2.8  | 2.6     | 2.4  | 2.2      | 2.4  | 2.2  | 2.8  | 2.6    | 2.4  | 4.2  | 4.4  | 4.4     | 4.2  | 4.6      | 4.8  | 4.2  | 4.6  | 4.2    | 4.4  |
| COD                     | ppm            | 68   | 64   | 66      | 64   | 62       | 62   | 60   | 62   | 67     | 68   | 42   | 44   | 40      | 42   | 40.0     | 38.0 | 40.0 | 42   | 42.0   | 42   |

BW: Bore well, OW: Open well